

MODULE

Histology and Cytology



Notes

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SPECIAL PROCESSING

17.1 INTRODUCTION

Some pathological specimens require special handling and need to be processed in a different way to reach the final diagnosis. Examples include eyeball, bones and bone marrow biopsy. The technical person needs to be aware of these special specimens so that appropriate measures can be taken before the grossing procedures are undertaken.



OBJECTIVES

After reading this lesson, you will be able to:

- explain the principle of special processing
- explain the different specimens requiring special processing
- learn how to do special processing procedures.

17.2 BONE

Normal human skeleton has two main types of bones: cortical or compact bone which is hard, solid and very strong and forms shafts of long bones i.e. the femur and tibia etc; and spongy or trabecular/ cancellous bone is found in the marrow cavities and is a mesh of bone strands which is almost ideal weight bearing structure particularly in the femoral head and vertebrae. The three major components of bone are mineral, cells and an organic extra-cellular matrix i.e. collagen fibers. The main bulk of bone is approximately 70% mineral and 30% organic components by weight. Bone cells are relatively few as opposed to marrow cells. The mineral of bone is mainly calcium and phosphate.

Techniques for the demonstration of bone and its components include:

Special Processing

- For decalcified bone: frozen, paraffin, or celloidin sections, transmission electron microscopy.
- For mineralized bone: frozen, plastic, scanning and transmission electron microscopy.

In order to obtain satisfactory paraffin sections of bone, inorganic calcium must be removed from the organic collagen matrix, calcified cartilage and surrounding tissues. This is called decalcification and is carried out by chemical agents, either with acids to form soluble calcium salts or with chelating agents that bind to calcium ions. Any acid, however well buffered, can have damaging effects on tissue staining. The problem increases with acidity of solution and duration of decalcification period. It mostly affects the nuclei which fail to take up hematoxylin and other basic dyes. These effects can be reduced by doing the decalcification end point test, post-decalcification acid removal and adjustment of the stain procedure.

17.3 DECALCIFYING AGENTS

Acids

Acid decalcifiers can be divided into two groups: strong (inorganic) and weak (organic) acids.

Strong Inorganic acids: e.g. nitric and hydrochloric acids may be used as simple aqueous solutions (5-10%). They decalcify rapidly but cause tissue swelling and can seriously damage tissue stainability if used longer than 24-48 hours. Old nitric acid is particularly damaging and should be replaced with fresh stock. They also damage tissue antigens for immunohistochemistry and enzymes may be totally lost. They can be used for small needle biopsies to permit rapid diagnosis within 24hrs. They can also be used for large or heavily mineralized bones with decalcification progress being carefully monitored.

Aqueous Nitric acid, 5-10%

Nitric acid	5-10 ml
Distilled water	To make 100ml

Formalin-nitric acid

Formaldehyde (37-40%)	10 ml
Distilled water	80 ml
Nitric acid	10 ml

Weak organic acids: e.g. formic, acetic and picric acid. Of these three formic acid is the only weak acid which is used in decalcification. Other two are used as components of other fixatives. Formic acid solutions can be aqueous (5-10%),

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buffered or combined with formalin. The formalin-10% formic acid mixture fixes and decalcifies simultaneously and can be used for small biopsies. Formic acid is suitable for most routine surgical specimens, particularly when immunohistochemistry is needed. Decalcification is usually complete in 1-10 days and decalcification progress should be monitored by a decalcification end point test.

Aqueous formic acid

90% stock formic acid	5-10 ml
Distilled water	To make 100 ml

Formic acid-formalin

90% stock formic acid	5-10 ml
Formaldehyde (37-40%)	5 ml
Distilled water	To make 100 ml

Buffered formic acid

20% aqueous sodium citrate	65 ml
90% stock formic acid	35 ml

17.4 CHELATING AGENTS

The chelating agent used for decalcification is ethylene-diaminetetracetic acid (EDTA). Although called an acid, it does not act like acids. EDTA will not bind to calcium below pH 3 and is faster at pH 7-7.4. This is a very slow process that does not damage tissues or their stainability, it also retains good antigen preservation for immunohistochemistry or enzyme staining and electron microscopy. The time required to totally decalcify dense cortical bone may be 6-8 weeks or longer although small bone spicules may be decalcified in less than a week.

Formalin-EDTA

EDTA, disodium salt	5.5 g
Distilled water	90 ml
Formaldehyde (37-40% stock)	10 ml

Aqueous EDTA, pH 7.0-7.4

EDTA, disodium salt	250 g
Distilled water	1750 ml

If solution is cloudy, adjust to pH 7 with sodium hydroxide.

Factors influencing the rate of decalcification

There are several factors influencing the rate of decalcification. The concentration and volume of the active reagent, including the temperature at which the reaction takes place, are important at all times. The more concentrated acids solutions decalcify bone more rapidly but are more harmful to the tissue. The usual recommended ratio of volume of decalcifying fluid to volume of tissue is 20:1 and the fluid should be changed several times during the decalcification

Process. Increased temperature accelerates decalcification, but it also increases the tissue damage and loss of heat sensitive antigen and enzymes. Other factors that contribute include the age of patient, type of bone, size of specimen and solution agitation. Mature cortical bone decalcifies more slowly than immature bone.

Treatment following decalcification

Acids can be removed from tissues or neutralized chemically after decalcification is complete. It can be done by immersing the bone into either saturated lithium carbonate solution or 5-10% aqueous sodium bicarbonate solution for several hours. Many laboratories rinse the specimens with running tap water for a period of time.

Eyeball

The eye should be placed in fixative as soon as practical after removal. Many of the tissues, the retina in particular, are very sensitive to anoxia and the longer you wait to fix the eye, the greater will be the artifacts, making interpretation difficult.

If eyes arrive in formalin and have been fixed for 2 days, wash them in water to remove the formalin (2 changes about 5 minutes each) and place them in enough 50% ethanol to cover the eye. Let the eye equilibrate overnight. Change the alcohol the next day and equilibrate for a second day. The eye should return to a normal volume and should not be indented or shrunken. For sectioning the eye it is best to wait 2 days with the eye in 50% ethanol.



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INTEXT QUESTIONS 17.1

1. For sectioning of the eyes, eyes must be placed in for atleast
2. Removal of inorganic calcium from the matrix is called

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3. Examples of strong inorganic acids are,
4. Examples of weak organic acids are, &
5. Chelating agent used for decalcification is



WHAT HAVE YOU LEARNT

- When eye balls placed in formalin are taken for sectioning, it is best to place the eye ball in 50% ethanol for 2 days before sectioning
- Gross description of the eye should be recorded, appropriate measurements of eye, tumor or pathologic processes also need to be recorded
- For decalcified bone techniques like frozen, paraffin, or celloidin sections, transmission electron microscopy are used
- For mineralized bone frozen, plastic, scanning and transmission electron microscopy techniques are used
- Inorganic calcium must be removed from the matrix by decalcification either by acids or chelating agents
- Acid decalcifiers are strong inorganic acids like nitric acid and hydrochloric acids and weak organic acids like formic, acetic, picric acids
- Chelating agent used for decalcification is ethylene diaminetetraacetic acid
- Several factors like concentration and volume of active reagent, temperature at which reaction takes place influence the rate of decalcification
- Acids from the tissues be removed from tissues or neutralized chemically after decalcification is complete by either saturated lithium carbonate solution or 5-10% aqueous sodium bicarbonate solution



TERMINAL QUESTIONS

1. Describe decalcification.
2. Enumerate the different methods of decalcification.
3. What are the advantages of EDTA decalcification?
4. What are the factors influencing the rate of decalcification?
5. Describe the special steps required before processing eyeball.



ANSWERS TO INTEXT QUESTIONS

17.1

1. 50% ethanol, 2 days
2. Decalcification
3. Nitric & Hydrochloric acid
4. Fromic, acetic & Picric acid
5. Ethylene-diaminetetraacetic acid (EDTA)

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